Properties of Natural and Simulated Diamonds

Gem	Source	Composition	Common dates of use	Mohs' hardness	Refractive index	Specific gravity	Dispersion	Crystal indices	Fluorescence	Optical	Other
Corundum	natural or synthetic	AI2O3	1900- 1947	9	epsilon = 1.757-1.768; omega = 1.765-1.776	3.96- 4.05	0.018	hexagonal crystal system with tabular, prismatic or pyramidal crystals; fracture = conchoidal	orange to red; heat treated stones may fluoresce green	marked dichroism	natural inclusions minerals and fluid are common; synthetic may have gas bubbles and curved striae; low thermal conductivity
Cubic zirconia	synthetic	ZrSiO4	1976 - present	8.5	2.15-2.18	5.6-6.0	0.06	fracture = conchoidal	yellow fluorescence in shortwave UV		usually no inclusions; low thermal conductivity
Diamond	natural	С	1476 - present	10	2.4175	3.51- 3.53	0.044	cubic crystal system; perfect cleavage in four directions.	may fluoresce pale colors in longwave UV; may phosphoresce		Numerous inclusions; trigonal on surface; very high thermal conductivity
Garnet	synthetic	YAG/GGG	1970- 1976	8.25/7.0	1.87/1.97	4.6/7.0	0.028/ 0.045	cubic crystal system		isotropic	turns brown in UV light; low thermal conductivity
Moissanite	synthetic	silicon carbide	1998 - present	8.5-8.25	2.65-2.69	3.22	0.104		may have pale fluorescence	strongly birefringent; anisotropic	may have brown tint; resistant to heat; inclusions occur as fine white tubes; high thermal conductivity
Paste diamond	synthetic	high lead glass	1700 - present	5.0-6.0	1.67	2.4-4.2	>0.02			highly refractive	low thermal conductivity; used since 1700
Quartz	natural	SiO2		7	epsilon = 1.553; omega = 1.544	2.65- 2.66		trigonal crystal system; low birefringence; fracture = conchoidal		low birefringence	heat treatments may bleach stones; low thermal conductivity; low thermal expansion;
Rutile	synthetic	TiO2	1947- 1955	6	2.6-2.9	4.25	0.33			doubly refractive	may have yellow tint; low thermal conductivity
Spinel	natural or synthetic	MgAI2O4	1920- 1947	8	1.715-1.725	3.5-4.1	0.02	isometric crystal system; fracture = conchoidal	natural fluoresces red in longwave; synthetic may fluoresce in shortwave	anisotropic	Natural stone has octahedral inclusions; may have fingerprint patterns; low thermal conductivity
Strontium titantate	synthetic	SrTiO3	1955- 1970	5.5	2.41	5.13	0.19				low thermal conductivity
Tourmaline	natural	aluminum borosilicate		7-7.5	epsilon = 1.610-1.650; omega = 1.635-1.675	2.9-3.2		hexagonal, prismatic crystals; fracture = conchoidal	little fluorescence	pleochroic; high birefringence	inclusions include gas or liquid pockets and color zoning; develops electrical charge when heated
Zircon	natural	ZrSiO4		6.0 - 7.5	epsilon = 1.968-2.015; omega = 1.923-1.960	4.6-4.7	0.039	tetragonal system with square prismatic crystals; brittle; fracture = uneven	some show dull yellow color; some may phosphoresce	pleochroic; high birefringence	heating brown zircon crystals produces strong colors (blue, green, red, etc.) that fade slowly with time or with UV exposure; low thermal conductivity